

# Density

**Density:** The amount of matter ("stuff") in a given volume (amount of space).  
To calculate the density of an object, you need to divide the mass of the object by its volume.

$$\text{Density} = \frac{M}{V}$$

Units of Density: g/cm<sup>3</sup>, g/mL, kg/m<sup>3</sup>, kg/L

**Mass:**

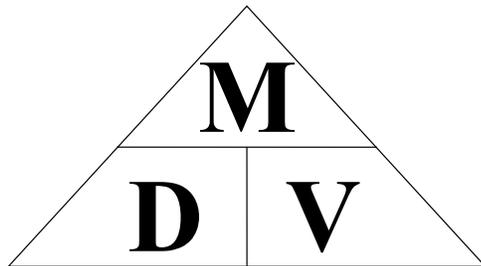
With many objects, mass can be measured using a balance or a scale. If the material is a liquid, you would need to subtract the mass of the container.

**Volume:**

To measure volume of a regular sized object measure the height, width and length of the object and multiply these three dimension. For a irregularly shaped object, use a graduated cylinder and find the displacement of the object.

**Formulas for Density, Mass and Volume:**

The triangle of mass, density and volume allows for you to solve for any one of these



Density =  $\frac{\text{mass}}{\text{volume}}$

$D = \frac{m}{v}$

Mass = density X volume

$M = d \times v$

Volume =  $\frac{\text{mass}}{\text{density}}$

$v = \frac{m}{d}$

**Density of Water = 1 g/cm<sup>3</sup>**

if an object's density < 1 g/cm<sup>3</sup>, it will **float**  
if an object's density > 1 g/cm<sup>3</sup>, it will **sink**

Fill in the blanks with the proper formula and then its proper number.

Place an X in the column if the substance will float or sink in water. Give proper units for each answer (g, cm<sup>3</sup>, g/cm<sup>3</sup>). Round answers to the tenths place.

<u>Formula</u>	<u>Mass</u>	<u>Volume</u>	<u>Density</u>	<u>Sink?</u>	<u>Float?</u>
	12 g	0.5 cm <sup>3</sup>			
	16 g		0.2 g/cm <sup>3</sup>		
		5 cm <sup>3</sup>	20 g/cm <sup>3</sup>		
	0.9 g		0.3 g/cm <sup>3</sup>		
	6 g	4 cm <sup>3</sup>			
	8 g	1.8 cm <sup>3</sup>			
		13 cm <sup>3</sup>	9 g/cm <sup>3</sup>		
	14.7 g	15 cm <sup>3</sup>			
		2 cm <sup>3</sup>	1.1 g/cm <sup>3</sup>		
	17 g		3 g/cm <sup>3</sup>		

## Problems:

Answer the following questions

1. How is the amount of space an object takes up measured?

By measuring its \_\_\_\_\_

2. What is the definition of density?

3. How do you find the density of an object?

4. What is a unit of density?

5. A baseball has a density of  $1.5 \text{ g/cm}^3$ . If you cut the ball in half, what will the density of one of the pieces be?

6. You know the volume of a glass of milk. What else do you need to know to find the density of the milk?

7. You have a brick with the dimensions of 2 cm by 3 cm by 8 cm. What is the volume of the brick?

8. A piece of wood has a mass of 18 grams and a volume of  $30 \text{ cm}^3$ . What is the density of the wood?

9. A graduated cylinder is filled to the 50 ml mark. When a marble is put into the graduated cylinder, the water level rises to the 54 ml mark. What is the volume of the marble?

10. A small rock has a density of  $1.4 \text{ g/cm}^3$  and a mass of 70 grams. What is its volume?